### LABORATORY

OF THE

# INLAND REVENUE DEPARTMENT

BULLETIN No. 90

HONEY, 1903

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### HONEY, 1903

Laboratory of the Inland Revenue Department, Ottawa, October 2, 1903.

W. J. Gerald, Esq., Deputy Minister of Inland Revenue.

Sir,—I beg to submit herewith a tabulated statement (No. I) giving a description of the samples of honey which were collected according to your instructions of March 27 last, and which have been submitted to examination in this laboratory. Before referring to the particulars given in the table, it seems necessary to make the following

explanatory remarks.

Honey is generally understood to be the sweet secretion which working bees produce from feeding on the nectar of the flowers, leaves, etc., of various plants and trees. As is well known, the aromatic constituents of many of these flowers are found in the honeys produced from them. Thus, in this country, clover, buckwheat and other honeys have been distinguished by their taste and aroma, while, on the continent of Europe, such names occur as linde, acacia, heath, conifer, forest and spruce honey. The nectar of flowers contains from 60 to 90 per cent of water, and both fruit sugar and cane sugar have been found in it. It experiences, in the stomach of the bee, certain changes which consist principally in an inversion of the cane-sugar. As is the case with many articles of food in Canada, 'the limits of variability' (see Section 19 of the Adulteration Act) permissible in honey have not yet been legally determined, but it seems to be generally accepted, not only by beekeepers but by the general public, that the feeding of bees in summer time with cane sugar or sugar solution, in order to increase the production, should be regarded as adulteration. This principle is expressly acted on by the association of Swiss Agricultural Chemists, who have also adopted 16 per cent cane sugar as the maximum limit which genuine honey ought to contain. Other chemists place the limit lower, and König states that natural honey may contain up to 8 or 10 per cent cane sugar. Experiments are on record which demonstrate that bees fed exclusively on cane sugar syrup produce a honey containing as much as 30 per cent cane sugar. Besides this adulteration, effected with the co-operation of the bees, honey may also be falsified by the direct addition of cane sugar or glucose (Starch sugar) syrup, the sophisticated honey in the latter case being sometimes sold as 'Swiss honey'. Neither of these varieties is difficult of detection, but it is otherwise in cases of an admixture of invert sugar, a substance produced by the action of dilute acids on cane sugar. This sort of adulteration has long been known, especially on the continent of Europe, where the product is sold under such names as 'Turkish honey', 'table honey' and 'prepared honey'. Not unfrequently it is more honestly termed 'sugar honey', or 'artificial honey,' (Kunst honig), and some of it is said to have come from America, in comb made from paraffine, and labelled as 'prime American honey.' Its manufacture has increased greatly during the last ten years. It is prepared with such skill as to baffle the efforts of German chemists to distinguish it from the genuine article, although both the German and the Belgian governments have sought by every means to protect the producers of pure honey. A recent memoir, emanating from the Imperial German Health Bureau, acknowledges that the trade in this artificial honey is very considerable, and that the addition of invert-sugar to genuine honey cannot be detected. (See Zeitschrift für Untersuchung der Nahrungs- u. Genussmittel for June 15, 1903.)

Whether it is likely that the last mentioned variety of spurious honey is sold in Canada, I am unable to state. I doubt whether invert-sugar is manufactured in Canada, and certainly it is not imported under that name. There is, however, a considerable importation of honey itself, which is given in the Trade and Navigation Tables as

follows:

Entered for Home Consumption for Fiscal Year ended

Honey in the comb, or otherwise,	June 30, 1902.	June 30, 190
and imitations thereof—	Lbs.	Lbs.
From Great Britain	31,856	6,209
" British West Indies	18,754	
" Austro-Hungary	6,571	E1 700
" China	320	51,789
" Germany	3,559	
" United States	85,451	62,606
Total	146,511	120,604

It would seem quite possible that some of this imported honey, upon which a duty of 3 cents per lb. has to be paid, may be of a spurious character.

The adulteration of honey by means of added water has also to be taken into consideration. The quantity in genuine honey seldom exceeds 20 per cent, and the maximum limit is not placed beyond 25 per cent by those chemists who have studied the subject.

In Table I will be found all the information, as well as analytical results, which is available regarding 99 samples of honey, which were collected in the open market and subjected to examination here. Besides the vendor's name and address, and those of the party from whom he is said to have procured the honey, there are given in a few cases designations indicating quality. Under the results of examination are given the direct saccharimeter reading, and the water percentage as the best discriminating tests for separating the apparently genuine from the apparently adulterated samples.

The direct saccharimeter reading of the 99 samples was observed in the same manner as in testing samples of cane sugar or molasses, by means of the Schmidt & Haensch improved instrument with triple field of vision. The number of degrees mentioned are therefore from observation of a 26.048 per cent solution in a 200 mm. tube. It will be observed that the great majority of the samples, 86 in number, give the levotatory readings, said to be indicative of pure honey when an admixture of invert-sugar is excluded from consideration. It must not, however, be supposed that left hand rotation, even in the absence of invert-sugar will infallibly indicate a genuine honey. It is quite possible that honey, adulterated with a small proportion of glucose syrup,

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may still give a distinct levo-rotatory reading. An experiment on this point, made by Mr. McGill, may here be put on record. 93 parts of honey reading —16 2° at 20° C. were mixed with 7 parts of ordinary glucose, reading +88° at 30° C. This adulterated honey read (after boiling and cooling the solution to correct bi-rotation) —4 · 3° at 20° C. The most of the samples, 69 in number, showed crystallization, while in 17 there was no such appearance. In this separation of the sugars, dextrose or grape sugar preponderates over the levulose or fruit sugar. These sugars are present in the honey itself in somewhat different proportions from those contained in invert-sugar, the levulose prevailing to a slight extent. Since 'it is held by experienced beekeepers that all genuine honey will eventually crystallize, and hence that honey warranted to remain syrupy is probably adulterated '(Allen), it is quite possible that some of these clear honeys may have contained an addition of invert-sugar. No less than 13 samples showed positive rotation, and, as probably adulterated samples, were subjected to further examination, the details of which are given in a separate table, No. II. The final conclusions arrived at regarding the adulteration of these are however incorporated in Table I, while the more technical explanations regarding the methods of examination are given in a memorandum appended to Table II.

The percentage of water stated in Table I was obtained by evaporating 10 ccm of a 5 p.c. solution of the sample in crysotile fibre for 24 hours at a temperature varying from 60° to 70° Centigrade, but never exceeding the latter figure. Only 8 of the samples were found to contain moisture to the extent of 25 p.c. and over, namely:—

No.	17887	with	27.4	p.c.	and		reading;	clear.
44	17896	"	25.6	* "			"	**
**	21300	44	25.0	66			"	66
66	21301	44	32.6	66			"	cryst.
66	21305	66	28.8	66	and	+	reading ;	clear.
66	23101	66.	26.4	64			"	"
46	17464	66	25.6	**	and	-	reading;	cryst.
"	17467	"	27.4	"	and	+	reading;	clear.

Significantly enough, three of these belong to the samples showing right-handed polarization and classed as adulterated. Other three belong to the class showing left-handed polarization, but suspected of adulteration with invert-sugar on account of their clearness. The remaining two gave minus readings and crystallization and are the only ones which may fairly be suspected of containing added or, at any rate, too much water.

It will be observed that some of the adulterated samples contained fragments of comb floating in the honey, which had to be separated previous to examination. This is, of course, a very different thing from honey in the comb. Most likely the general consumer might avoid adulterated honey by purchasing it in the comb and carrying out himself the process of extracting the boney.

According to the conclusions stated in Table I, the following classification may be made of the samples described in it:-

Genuine	81
Doubtful	
Adulterated	
Adulterated with glucose syrup	6
Adulterated with cane sugar	5
Total number of samples	. 99

I have the honour to be, sir, your obedient servant,

> THOMAS MACFARLANE, Chief Analyst.

lection.		Sample.	NAME AND	Address of				Brand
Date of Collection.		Number of	Vendor.	Manufacturer or Furnisher as given by Vendor or on the Label.		ntity cted.	Cost.	Or Description.
1903			District of Halifax.				\$ cts.	
April	16	20240	R. F. Guest, Yarmouth	Parker & Eakins, Yarmouth	3 bot	tles	0 30	
	20	20263	R. B. Dakin, Windsor,	Brown & Webb, Halifax	3		0 20	
	21	20264	H. E. Wilson, Windsor	Hattie & Mylius "	3		0 20	
"	23	20312	G. A. Burbridge, Halifax.	Brown & Webb "	3		0 30	
11	28	20313	Hattie & Mylius "	Vendors	3		0 45	
11	30	20322	R. McFatridge "	Hattie & Mylius, Halifax	3		0 45	'Fine N w
**	30	20325	W. H. Stevens, Dartmouth		3		0 30	Honey.'
"	30	20328	E. Butcher "	Brown & Webb "	3		0 45	
May	1	20330	G. H. Caldwell, Halifax		3		0 40	
"	1	20333	Brown Bros. & Co. "	" "	3		0 45	
Apri	117	4317	G. E. Hughes, Charlotte-		3		0 45	
*	17	4318	town. H. A. Ellis, Charlottetown	John Newson, Charlotte	3		0 38	
**	17	4327	G. A. Gourlie, Summerside	Henry Watson & Co	3		0 45	
	17	4334	D. Gordon, Georgetown	Evans & Son, Montreal	3		0 45	
	17	4340	A. McLean "	Dearborn & Co	3		0 60	
			District of New Brunswick.					
"	9	17854	Puddington & Merritt, 55 Charlotte St., St. John	E. L. Colpitt & Co., Petit codiac, Pleasant Vale	3	"	0 75	'Maple Leaf'.
	11	17856	G. M. & A. A. Barker, 100	Albert Co., N.B. R. H. Smith, St. Thomas.	3		0 45	
**	14	17867		Bottled by vendor from	13		0 60	Honey.
	17	1788	St., St. John. Francis McKay, 228 Main	honey in bulk. F. W. Fearman, Hamilton	3		0 45	'Clover Honey'
"	17	17886	Dr. E. O. Steves, 301 Mair	Canadian Drug Co., St	. 3		0 45	
	17	17887	St., Moneton. Geo. Spencer, 272 Mair		3		0 35	'White Clover Honey.'
"	17	17888	St., Moncton. S.J. MeD. Cook, 195 Main	W. D. Black, Cloverdal	е 3	"	0 75	'Extracted CloverHoney'
"	21	1789	St., Moncton. Fred Waterson, 4 King St.	Apiary. Truro, N.S. Canadian Drug Co., St	. 3		0 60	'Clover Honey
**	22	1790	St. Stephen.	John. G. F. Beach, Charlotte Co. N.B.	, 3		0 60	
"	25	1791		B. A. Goodspeed, York	1		0 75	
			District of Quebec.				lain.	
"	16	2333	9 C. Peloquin, Notre Dame de St. Hyacinthe.	Vendor	. 3	lbs	0 30	

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Direct Sacchari- meter Reading.	Water.	Physical Characters.	Observer.		the	Number of Sample.	
							-
	р. с.						
- 9.0	24.6	Light yellow syrup, clear with M	Iiss S. E. Wı	right.	Jenuir	ne	2024
-7.3	24.0	Thick; strong beeswax smell, crystallised throughout.	"		"		2026
- 7.8	24.4	Clear syrup with sediment, strong	"		"		2026
-32.2	17.0	beeswax smell. Thick; crystallised throughout,	"		"		2031
- 2.3	22.2	strong beeswax smell.  Deep yellow syrup, clear; bees-	"				2031
-10.7	18.8	wax smell; mild taste.  Deep yellow colour; clear; mild	"		"		2032
- 7.9	14.0	flavour. Somewhat thick; slightly smoky	"				2032
-10.2	22.6	odour. Yellow colour; one-half clear; theothercrystallised; mild taste;	"		11		2032
-10.7	19.2	beeswax odour. Thick crystallised syrup; mild	"		**		2033
-14.2	18.4	taste.	"		11		2033
-14.3	18.2	Light yellow colour ; crystallised M	Iiss E. Davie	dson.	"		431
- 9.0	18.0	throughout. Yellow; slightly crystallised			"		431
-15.7	18.8	Lightyellow; crystallized through-	,,		"		432
-15.4	23 8	out. Yellow; slightly granulated or			"		433
+26.4	22.4	crystallised. Yellow; partly crystallised	"			erated by addition ose syrup.	434
-16.9	22.0	Brown; partly granulated	"		Genuir	ne	178
-15.0	19.4	Yellow; granulated throughout.	,,		"		178
- 9.0	20.6	" clear	.,		"		1786
-17:0	21.6	" partly granulated	"		"		1788
-18.3	18.2	" " "			h		1788
-15.7	27:4	Light yellow; clear; pleasant	diss S. E. W	right.	Doubt	ful	178
-10.4	25.4	taste. Light yellow; deposit on bottom	"	-	Genui		178
-13.3	25.6	and sides of jar; mild.			Doubt		178
-15.8	24.6	taste.	"		Genui		179
-15.6	23.2	mild taste.	"		"		179
- 8.7	18.2	Bright yellow colour; clear; flower-	"		"		233

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-14 -16 -15 -16 -16 -15 -18 -13 +83

- 8:9
-11:1
-13:9
-15:1
-14:2
- 8:8
- 7:9

 $-17.5 \\ -13.9 \\ -11.2$ 

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lection.		Sample.	NAME AND	Address of	Quantity		Brand	
Date of Collection		Number of	Vendor.	Manufacturer or Furnisher as given by Vendor or on the Label.	Collected.	Cost.	Description.	
1903			District of Quebec—Con.			8 cts.		
April	22	23362	W.Campbell,St.Hyacinthe	W. Myers, White's Station	3 lbs	0.30		
	28	23372		Jos. Bissonnet, Valcourt	21 "	0 30		
"	29	23378		Le Rochefort, Bécancourt.	31 "	0 39		
May	1	23388		From farmer of Lachinail.	5	0 60		
April	30	23381	bonne. E. Goudron, Berthierville.	Vendor	3 "	0 30	White	
"	30	23382	" "		3 "	0 30	Brown	
May	7	23391	Simeon Papillau, Notre	Antoine Marcil, Notre Dame	25	0 23	****** *** **	
11	15	23604	J. B. St. Pierre, St. Hya-	de St. Hyacinthe. Sœurs du Precieux Sang.,	41 "	0 65		
"	15		cinthe.	St. Hyacinthe. C. Peloquin, Notre Dame de St. Hyacinthe.		0 30		
			District of Montreal.					
May	5	21296	H. Poirier, 1978 St. Cathe-		1 lb	0 12		
"	5	21297	P. Daoust, 1830 St. Cathe-	Gunn, Langlois & Co	1	0 12	************	
11	5	21298	A. Fournier, 1879 St. Ca-		1 "	0 12		
11	5	21299	therine St., Montreal. G. De LaMothe, 1502 St.		1	0 12		
11	5	21300	P. Massicotte & Co., 1470	John Miller, Montreal	2 jars	0 20		
11	8		St. Catherine St. Mont'l.			0 13		
	14	21302	Dame St., Montreal. L. P. Lavoie, 3187 Notre	Hudon & Orsali, Montreal.	1	0 10		
11	14	21303	Dame St., St. Cuneg'de. Robert & Frères, 229 Rich-	Hudon & Orsali, Montreal.	1 "	0 12		
11	14		L. Legault, 102 Coursol St.	L. P. Lavoie, St. Henri		0 10		
11	14	21305	St. Cunégonde. W. J. Maloney, 468 St. Antoine St., St. Cunégonde.	Montreal Canning and Pre- serving Co.	3 jars	0 30	'Fine Honey Compound,'	
			District of Kingston.				'Banner Brand.'	
April	30	23137	S. Fourt, Walton St., Port		1 bottle	0 20		
"	30		Hope. D. F. McDonald, Pitt St.,	W. Atchison, Cornwall		0 27		
"	30		Cornwall. W. A. Dunkin, Cornwall.			0 15		
"	30			J. McNaughton, St. Ra-	THE REAL PROPERTY.	0 10		
"	30	23142	D 37.1	phaels. C. C. Farran, Farran's Pt.	3 jars	0 40		
"	30 30	23143 23144			3 " 1 lb	0 30 0 11		
"	30					0 12		
"	30	20146	St., Cornwall.		1 "			
"	30	23138	J. Maybury & Co., Prescott.	Mrs. Robt. Johnson				

Direct Sacchari- meter	RESUL	Physical Characters.	Obs	server.	Remarks by the Chief Analyst.	Numiber of Sample.
Reading.						N
	p. c.					1
-13.7	23.4	Brownish yellow; seven-eights of the bulk is granulated; smoky	Miss S.	E. Wrigh	nt Genuine	23362
- 6.2	22.8	smell and taste. Yellow; clear; very strong bees- wax smell.		31 ·		25372
-12.2	22.6	Brown; clear; buckwheat taste and smell.		" .		23378
-11.5	22.6	Light yellow syrup; thick; strong beeswax smell; mild taste.		11	"	23388
$-12\ 5$	20.0	Light brown colour; partly crystallised.	Miss E.	Davidsor	1. "	23381
-16.6	17.8	Brown; crystallised throughout; buckwheat taste.		"	"	23382
-15.7	19.2	Light yellow; crystallised through- out; smoky smell.				23391
- 5.8	19.0	Light yellow; crystallised throughout.		,,		23604
-16.0	19.2	Yellow; crystallised almost throughout.		"		23605
-14.3	20.0	Light yellow colour: crystallised	Miss E.	Davidsor	a. Genuine	21296
-16.5	19 2	throughout. Light yellow colour; crystallised				21297
-15.5	19.2	Yellow; crystallised throughout;				21298
-16.7	20.0	slight odour of smoke.  Light yellow; almost entirely				21299
-16.6	25 0	crystallised. Light yellow, clear; very sweet			Doubtful	21300
- 6.0	32 6	Brown; partly crystallised; tastes		" .	Contains too much water	21301
-15.8	21.0	of buckwheat. Light yellow; crystallised throughout.			Genuine,	21302
-18.5	20.8	Dark brown; crystallised; smells and tastes of buckwheat.		11	" •	21303
-13.8	24.4	Dark brown; deposit at bottom;	Miss S.	E. Wrigh	t. "	21304
+83.0	28.8	smells and tastes of beeswax.  Clear thin syrup; mild odour and not much taste of honey.		"	Adulterated by addition of glucose syrup.	21305
- 8.9	18.4		Miss S.	E. Wrigh	t. Genuine	23137
-11.1	14.6	like flavour. Light yellow; granulated; plea-		"	"	23139
-13:9	20.6	sant taste and smell.  Light yellow; granulated; mild flavour.		"		23140
-15.1	19.2	navour,		11		23141
$-14.2 \\ -8.8 \\ -7.9$	19·0 20·0 23·8	Yellow; clear; mild taste & smell. "granulated Pale yellow; thick syrup; mild		**		. 23143
$-17.5 \\ -13.9$	20·0 19·0	taste. Yellow; partly crystallised r crystallised throughout.	Miss E.	Davidson	n. "	. 23145 . 23146
-11.2	24.2	" strong smell and taste of beeswax.	Miss S.	E. Wrigh		00100

### RESULTS of Analysis of 99 Samples

lection.	Sample.	NAME AND	Address of	0		Brand
Date of Collection.	Number of	Vendor.	Manufacturer or Furnisher as given by Vendor or on the Label.	Quantity Collected.	Cost.	Description.
1903.		District of Toronto.			8 ets.	
April 15	23401	Chas. Chown, 575 Yonge St., Toronto.	Jones Bros., Niagara, Ont.	3 jars	0 27	
" 15	23402	Toronto		0 "		
15		Toronto.	T. J. Dougall, Stouffville, Ont.		0 15	
16		St. Barrie.	J. Gough, Crown Hill		0 10	
" 16		Barrie.	W. I. Goods, Cooper Will		0 13	
" 16		Barrie.	W. J. Gough, Crown Hill.		0 40	
18		St. Catharines.	H. Freas, St. Annes		0 30	
18	23101	St. Catharines.	Upton Co., Hamilton	3 jars	0 00	
17	23407	John O. Carpenter, M'ket		1 lb	0 20	
17	23408	Square, Hamilton. C. H. Peebles, Market Square, Hamilton.		1 "	0 15	
		District of Windsor.				
April 9	22047 22065	C. A. Nairn, Goderich Stuebing Bros., Berlin	Mrs. Strachan, Goderich F. F. Dalley & Co., Hamil-	3 jars 3 "	0 40 0 30	
. 16	22069	A. K. Roesch, Waterloo.	F. W. Fearman, Hamilton,	1 lb 3 jars	0 25 0 35	
" 18 " 22	22080 2 22087	J. A. McRae, Gueiph Wm. Anderson, Chatham.	ton. F. W. Fearman, Hamilton. Mrs. Rose Miller H. Walker & Sons, Guelph. Mr. Chrysler	3 " 3 lbs	9 30 0 35	
" 25 " 25	22088	Hugh Malcomson "	Amos Kelly, Co. Kent, Ont.	3 " 3 pots	0 35 0 30	
. 22	22097	James Wilson, London	Amos Kelly, Co. Kent, Ont. Wall & Guffy, Windsor E. Bainard, Glenworth	3 "	0 35	
25	2 22098	A. P. Yeo, London	Vendor	3 "	0 30	
		District of Winnipeg.				
April 18	17430		Dundas & Flavelle, Lindsay,	1 lb	0 60	
18	8 17438	den. J. A. Munro, Boissevain.	Ont. Dęadmaris, Brusselş, Ont.		0 75	
. 2	3 1744		H. H. Waddell, St. Thomas,	3 jars	1 05	
" 2 " 2	4 17445 5 1745	somin. 9 A. Grant, Brandon 9 J. & E. Brown, Portage la	Ont. From a farmer Bright&Johnson,Winnipeg	3 "	1 05 0 75	
May		Prairie.	Dyson, Gibson Co., Winni		0 60	
		J. G. Hargrave "	Upton, Hamilton		0 75	
	7 1746	9 Porter & Orris "	Dundas& Flavelle, Lindsay	3	0 60	
		2 Campbell Bros. & Wilson	Ont. Vendors	3 "	0 60	

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	RESUL	TS OF EXAMINATION.			D	Sample
Direct Sacchari- meter Reading.	Water.	Physical Characters.	Obs	erver.	Remarks by the Chief Analyst.	Number of Sample.
0	р. с.					
-15.9	21.4	Yellow; partly crystallised	Miss E.	Davidson.	Genuine	234
-17.7	22.8	Light yellow, crystallised almost			"	234
-15.7	19.0	throughout. White; crystallised throughout			"	234
- 7.4	18.4	and almost solid. Yellow; slightly crystallised				234
- 1.9	22.2	Yellow; partly crystallised				23-
-10.1	21.0	Light yellow; crystallised through-				234
16.2	21.6	out. White; crystallised throughout.	1000		"	23
+70.8	26.4	Yellow; clear			Adulterated by addition of glucose syrup and	233
-11.4	22.4	White; crystallised throughout	M. Sais		Genuine	23
-14.2	20.6				"	23
-19.2	21.6	Yellow; partly crystallised	Miss F	Davidson	Genuine	22
-19.2 $-17.0$	20.8	" crystallised throughout	MIISS EA	" ·	"	22
-7:0	19.2	" clear partly crystallised		"		22 22
$-14.1 \\ -20.0$	21·4 21·2			,,		22 22
-13.5	19.4	Light yellow; crystallised through out.	1	" .		22
$-13.0 \\ -12.0$	18.8	Yellow; partly crystallised		" .		22
-14.8	21 4	Light yellow; crystallised through out.		" .		22
- 9.1	20.2	Yellow; crystallised throughout		"	. "	22
-14.5	21.2	Yellow; partly crystallised	Miss E.	Davidson	Genuine	17
+23.0	23.2	Dark yellow; clear		н .	Adulterated by addition of cane sugar.	17
-15.3	19.2	Yellow; crystallised throughout			Genuine	17
$-7.4 \\ +19.0$	19·2 23·0			" :	Adulterated by cane sugar	17
-11:6	25.6				. Contains too much water.	17
+49.1	27.4	Dark yellow; clear			Adulterated by addition of glucose syrup.	17
-14.5	20.2	White; crystallised throughout.			Genuine	17
+26.4	16.4	Brown; clear; very thick			Adulterated by addition of cane sugar.	17

#### RESULTS of Analysis of 99 Samples

lection.		Sample.	NAME AND	Address of						Brand
Date of Collection.		Number of	Vendor.	Manufacturer or Furnisher or given by Vendor or on the Label.			tity ted.	Co	st.	or Description.
190	3.		District of Manitoba.					8	cts.	
Apri	21	21711		J. Turner & Co., Hamilton	3 ј	ars.		0	75	
11	23	21715	Deer. John West, Wetaskiwin		3	"		0	60	
"	24	21719		Dyson, Gibson Co., Winni-	3	"		0	60	
11	25 29		monton. A. Davies, Strathcona A. W. Ward, Calgary						75 60	
			District of British Columbia							
Apri	1 16	21686	A. T. Charleton, Port Haney.	Geo. Charleton, Ailsa Craig,	2	11		0	40	
11	16	21690	Coulter & Berry, Langley.	H. L. Johnson, Chilliwack	3	11		0	75	
"	16	21699	H. C. Henderson, Chilliwack.	A. Malcomson, Chilliwack.	3	"		0	75	
"	16	23503	Mrs. E. A. Farrer, Chilliwack.	G. A. Kipp, Chilliwack	3	"		0	75	
"	21	23516	S. Petersky, Steveston		3	11		0	90	
"	21	23520	E. Hunt, Steveston	San Diego Honey Co., San	3	11		0	90	
11	21	23522		Francisco. Pacific Const. Syrup Co.,	3	**		1	00	
11	21	23523		San Francisco. Schilling Co., San Francisco	3	11		0	90	
11	25	23542	Landing. Speed Bros., Victoria	J. Reagh, Ladners's Land-	3	"		0	75	
"	28	23549	E. J. Rae, New Westmin- ster.	w. Ross, Ontario Co., Cali- fornia.	3	"		0	75	•

#### of Honey as sold in 1903—Concluded

	RESUL	TS OF EXAMINATION.	Observer.			Sample.
Direct Sacchari- meter Reading.	Water.	Physical Characters.			Observer.	
	10 Ug					
-14.5	16.8	White; crystallised throughout	Miss E. David	son.	Génuine	21711
-11.0	18.8	White; crystallised almost	"			21715
-12.4	23.2	Yellow; partly crystallised	"			21719
+ 7·1 + 3·5	25·2 26·2	Yellow; clear	"		Adulterated by cane sugar	21726 21729
-17:4	16.6	White; crystallised throughout			Genuine	21686
- 8.4	22.8	Yellow; partly crystallised; con-	"			21690
+14.2	18.8	tains pieces of comb.  Brown; partly crystallised; very thick; tastes and smells of			Adulterated	21699
- 8.2	19.0	brown sugar.  Dark yellow or brown; crystal- lised almost throughout; con-			Genuine	23503
+69.8	21.6	tains pieces of comb. Yellow; partly crystallised; contains pieces of comb.	11		Adulterated with glucose	23516
+41.6	21 2	u u u	"		syrap.	23520
+42.8	21.0	Yellow; slightly crystallised;	"			23522
-14.8	17.0	contains pieces of comb. White; crystallised throughout.	"		Genuine	23523
-10.7	17:0	Yellow; partly crystallised				23543
-16.8	15.0	Brown; not crystallised; very thick.	"			23549

Table II.—Results of further examination of 13 samples Honey showing right handed rotation, by Miss. E. Davidson.

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tion chle 10 ther by 4 deri tain ence be s suffi estir cent samı of Ta is us manı per c gentl plane on m tion. to the

	В	Y CLERGE	T PROCES	88.	Approxi-	Br Fr	EHLING SOL	UTION.		
No. of samples.	Direct Saccha- rimeter reading.	Saccha- after rimeter Inver-		Cane sugar by Clerget formula; p. cent.	mate percentage of glucose syrup.	Reducing sugar, stated as invert.	Reducing sugar after inversion.	Cane sugar, per cent.	Dextrine	
4340	+27.4	+21.3°	23°	4.65	13.7	62.68	68.04	5.09	Distinct.	
21305	+81.0	+71.3	24	7.42	40.5	49.48	54.24	4.52		
23101	+70.7	+39.3	24	24.02	35.3	40.15	62.60	21.33	"	
17438	+22.9	-17.9	24	31.22		42.98	80.08	35.26	None.	
17459	+19.2	-16:5	24	27:31		47.15	76.80	28.16		
17467	+48.6	+35.2	24	10.24	24.3	52.98	63.00	9.52	Distinct.	
17472	+26.8	-16.4	24	33.05		45.69	78.75	31 41	None.	
21726	+8.2	-15.2	24	18.13		52.49	75.00	21.38	11	
21729	+3.9	-14.7	23	14.18		55.60	75.04	18.46	11	
21699	+14.1	+10.8	21	2.49		65.13	68.63	3.32	"	
23516	+70.1	+55.4	23	11.20	35.0	55.51	65.72	9.70	Distinct.	
23520	+41.9	+35.4	23	4.96	20:9	63.38	69 · 22	5.24		
23522	+43.5	+32.2	23	8.61	21.7	62.60	70.00	7.11	"	

#### MEMORANDUM REGARDING THE EXAMINATIONS REFERRED TO IN TABLE II.

The samples whose numbers are given in this table are those which shewed right handed rotation in the polariscopic observation noted in Table I, and which were subjected to further examination in order to ascertain whether this behavior was due to glucose syrup or cane sugar, and, in the latter case to determine the quantity of cane sugar present.

They were first examined by the Clerget process the nature of which is very clearly described by Allen (Commercial Organic Analysis, 1898; Vol. I, p. 260). The only difference which has been made in the equations there given is in the change by inversion which instead of 144 has been placed at 142.7 in accordance with the more recent determinations of Wohl. In Table II all the observations were given which are necessary for calculating the cane sugar, the percentage of which is also stated.

The percentage of sucrose present in the samples of Table II was also ascertained by the use of Fehling solution, the details of the process being as follows:—A five percent solution of the honey sample was first prepared.

(1.) For determining the reducing sugars 10 c.c.m. of it, containing 0.5 grammes of the original sample were treated direct with Fehling solution in excess. The weight of the cuprous oxide produced multiplied by the factor 0.4861 and by 200 gave the percentage of reducing sugars present, stated as invert sugar. The name reducing sugar applies to all the varieties of this substance which act upon Fehling solution with precipitation of cuprous oxide. Dextrose, grape sugar, starch sugar, levulose or fruit sugar, the mixture of dextrose and levulose called invert sugar, and certain reducing substances which form in the syrup during the manufacture of sugar from the cane are all included under "reducing sugars." The term excludes cane sugar which does not act upon Fehl-

ing solution previous to inversion. The factor 0.4861 is based upon work done by Mr. F. W. Babington (Analyst, Vol. xvi., p. 181) and represents the cuprous oxide yielded by 1 gramme of invert sugar derived from inverting by dilute acid 0.95 grammes of cane

sugar and precipitating by Fehling solution.

(2.) For ascertaining the quantity of cane sugar present 50 cem. of the above mentioned five per cent solution of the honey were inverted by the action of 2 ccm. hydrochloric acid, then rendered slightly alkaline by potash solution and made up to 100 ccm. 10 ccm. of this inverted solution, containing 0·25 grammes of the original sample were then treated with excess of Fehling solution. The cuprous oxide produced multiplied by 400 and the factor 0·4861 gave the percentage of reducing sugar, including that derived from the inversion of the cane sugar present. The direct percentage, as ascertained under (1) was then deducted from the percentage after inversion, and the difference multiplied by 0·95 which gave the percentage of cane sugar in the sample. It will be seen from the table that the percentages thus obtained confirm, on the whole, with sufficient accuracy, those obtained by the Clerget process.

Allen states that, in the absence of added cane and invert sugar, an approximate estimation of the proportion of glucose syrup in honey may be made by reckoning 1 per cent of the adulterant for every degree of dextro-rotatory power possessed by the sample. Following this rule the percentage of glucose syrup present in the 13 samples

of Table II are given in one of its columns.

The samples in question were also subjected to a qualitative test for dextrine which is usually a constituent of commercial glucose syrup. This test was applied in the manner described by Haenle (Die Chemie des Honigs; Strasburg, 1892), 5 ccm. of a 33½ per cent solution of the sample are placed in a test tube, and 2 ccm. of absolute alcohol gently added. If dextrine is present a white turbidity is observable at the contact plane of the two fluids, which is caused by the separation of the dextrine, and disappears on mixing. Cane sugar treated in this way, and honey as a rule do not give this reaction. In Table II a column is given which shows the results of this test when applied to the different samples.

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