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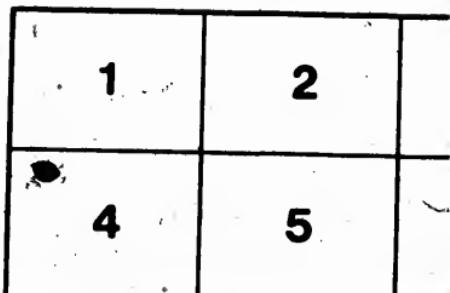
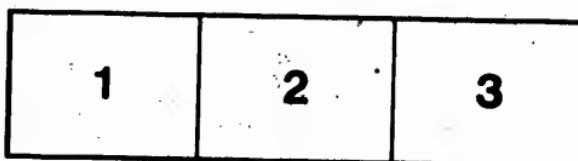
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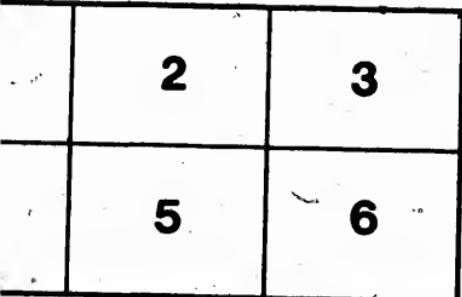
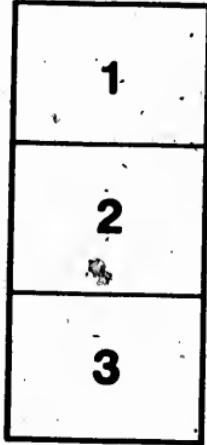
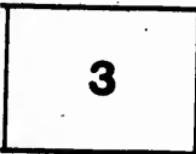
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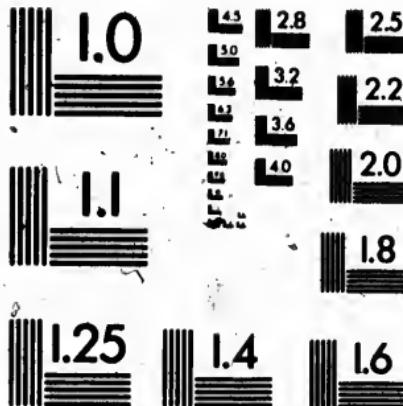
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Nov - 1839

TABLE

COMBINING WEIGHTS:

FOR

PROFESSOR CHAPMAN'S

COURSE ON MINERALOGY.

University College, Toronto.

1839.

*In the compilation of this Table, those substances only have been selected, which have a direct bearing on Mineralogical Science. The first column contains the name of the substance; the second, its symbol or formula; and the third, its combining weight. In the case of compound bodies, the percentage amount of the Electro-negative element is also given in a fourth column.*

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54.1.25  
454

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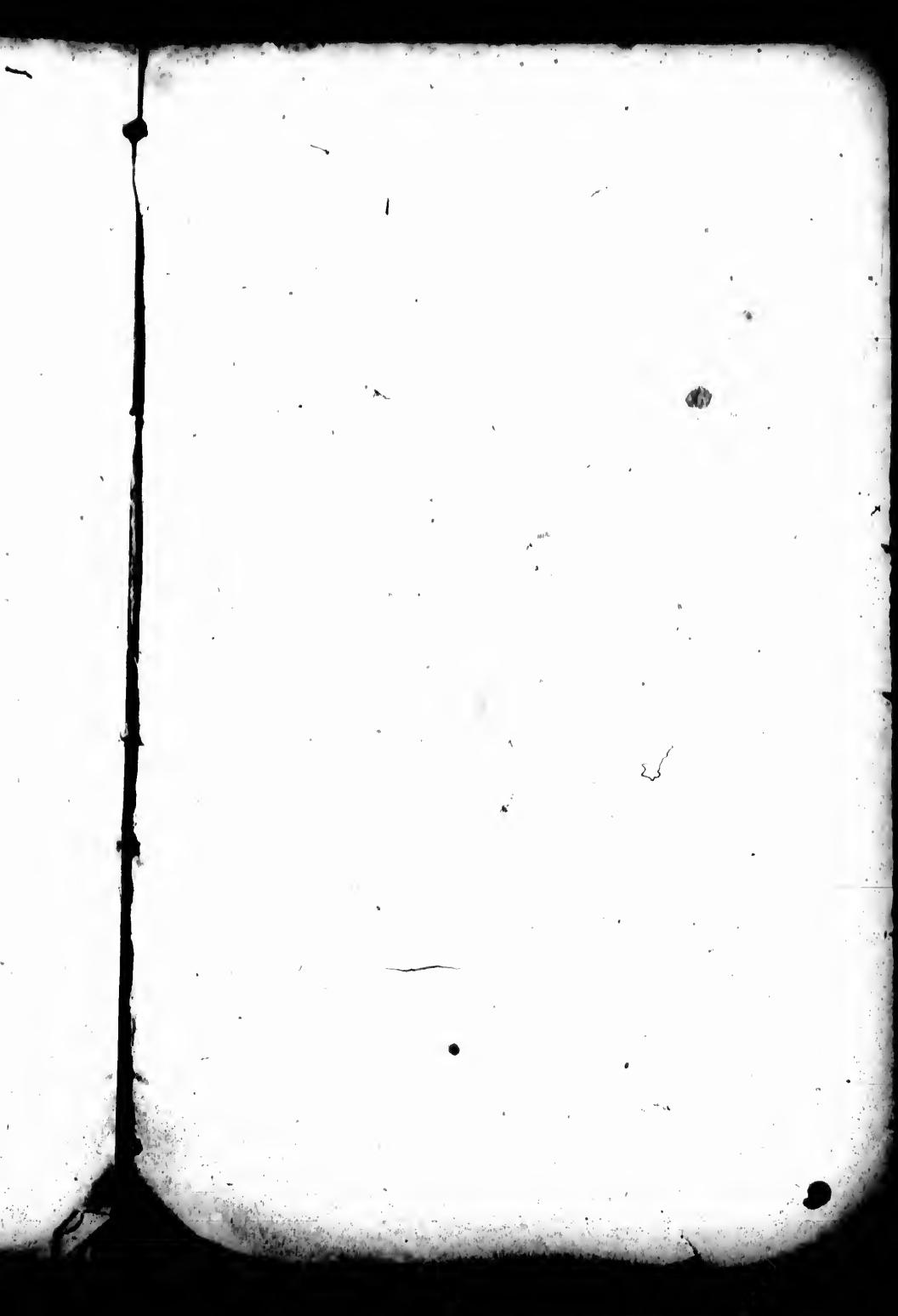
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COMBINING WEIGHTS.

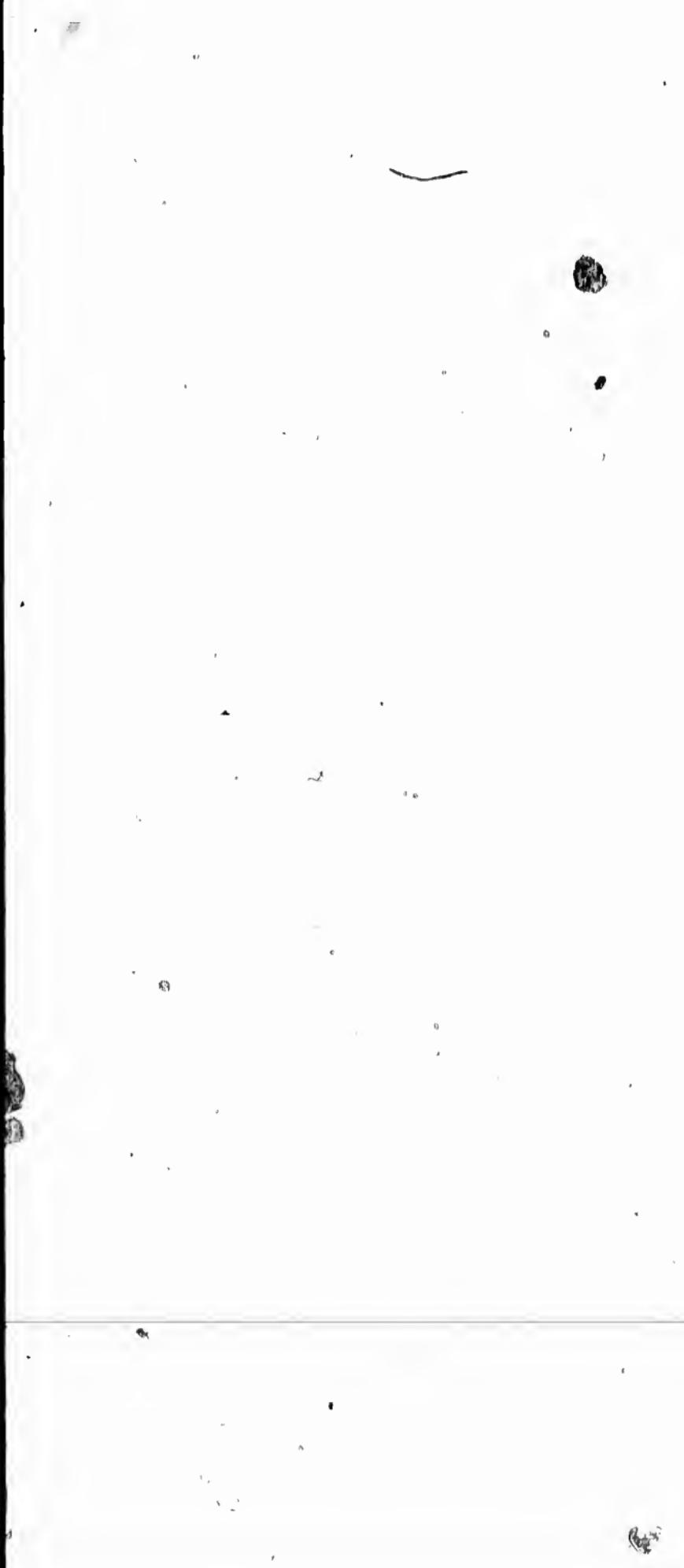
NAME OF SUBSTANCE.	FORMULA.	WEIGHT.	PER-CENTAGE. E-N. E.L.
ALUMINUM.....	Al	170.9	
Alumina .....	Al <sup>2</sup> O <sup>3</sup>	641.8	46.74
ANTIMONY (Stibium)....	Sb	1612.0	
Sulphide of Ant.....	Sb S <sup>3</sup>	215.15	27.19
Antimonius Acid .....	Sb O <sup>3</sup>	1012.0	15.68
Antimonic Acid .....	Sb O <sup>5</sup>	2112.9	23.66
ARSENIC ....9.....	As	937.5	
Sulphides of arsenic {	As S <sup>3</sup>	1530.75	39.11
Arsenious Acid .....	As S <sup>5</sup>	1041.25	51.71
Arsenic Acid .....	As O <sup>3</sup>	1237.5	24.24
As O <sup>5</sup>		1437.5	34.78
BARIUM.....	Ba	856.78	
Baryta.....	Ba O	956.78	10.45
BISMUTH.....	Bi	2660.76	
Sulphido .....	Bi S <sup>3</sup>	3263.0	18.46
Oxide .....	Bi O <sup>3</sup>	2900.76	10.13
BORON.....	B	186.2	
Boracic Acid .....	BO <sup>3</sup>	436.2	68.78
BROMINE.....	Br	999.62	
CADMUM .....	Cd	696.77	
Sulphide .....	Cd S	897.52	22.37
Oxide .....	Cd O	796.77	12.55
CALCIUM.....	Ca	251.65	
Lime .....	Ca O	351.65	28.44
CARBON.....	C	75	
Carbonic Acid.....	CO <sup>2</sup>	275	72.73
CERIUM.....	Ce	590.8	
Oxide .....	Ce O	690.8	14.48
Sesquioxide .....	Ce <sup>2</sup> O <sup>3</sup>	1481.6	20.52

NAME OF SUBSTANCE.	FORMULA.	WEIGHT.	PER-CENTAGE. E-N. E-E.
CHLORINE .....	Cl	443.28	
CHROMIUM .....	Cr	328.59	
Sesquioxide .....	Cr <sup>2</sup> O <sup>3</sup>	957.18	31.34
Chromic Acid.....	Cr O <sup>4</sup>	628.59	47.73
COBALT .....	Co	368.65	
Sulphides .....	{ Co S Co <sup>2</sup> S <sup>3</sup> Co S <sup>2</sup>	{ 569.4 1339.53 770.15	{ 35.26 44.96 52.43
Oxido .....	Co O	408.65	21.34
Sesquioxide .....	Co <sup>2</sup> O <sup>3</sup>	1037.30	28.92
COPPER (Cuprum)....	Cu	395.6	
Sulphido .....	Cu <sup>2</sup> S	991.95	20.24
Sub-oxide .....	Cu <sup>2</sup> O	891.20	11.22
Oxide .....	Cu O	495.6	20.18
DIDYMUM .....	Di	620	
Oxide .....	Di O	720	13.0
ERMIUM .....	E		
FLUORINE .....	Fl	235.43 [237.5?]	
GLUCINUM (Beryllium)	Be	87.12	
Glucina .....	{ or Be <sup>2</sup> O <sup>3</sup> or Be O	{ 58.08 474.24 158.08	{ 63.26
GOLD (Aurum) .....	Au	2458.33	
HYDROGEN .....	H	12.5	
Water .....	HO	112.5	88.80
IODINE .....	I	1586.0	
IRIDIUM .....	Ir	1232.08	
IRON (Ferrum) .....	Fe	350.52	
Sulphides .....	{ Fe S Fe <sup>2</sup> S <sup>3</sup> Fe S <sup>2</sup>	{ 551.28 1303.31 752.02	{ 36.41 46.21 53.39
Oxide .....	Fe O	450.52	22.20
Sesquioxide .....	Fe <sup>2</sup> O <sup>3</sup>	1001.05	20.07
LANTHANUM .....	Ln	588.0	
Oxido .....	Ln O	688.0	14.54
LEAD (Plumbum) .....	Pb	1294.64	
Sulphide .....	Pb S	1495.39	13.43
Oxide .....	Pb O	1394.64	7.17
Chloride .....	Bb Cl	1787.92	25.51
LITHIUM .....	Li	82.03	
Lithia .....	Li O	182.03	54.98

NAME OF SUBSTANCE.	FORMULA.	WEIGHT	PERCENTAGE E-N. E.
MAGNESIUM .....	Mg	150.19	
Magnesia .....	Mg O	250.19	39.07
MANGANESE .....	Mn	311.68	
Sulphides .....	Mn S	515.13	36.81
	Mn <sup>2</sup> S <sup>3</sup>	1291.62	46.62
	Mn <sup>2</sup> S <sup>3</sup>	746.18	33.81
Oxides .....	Mn O	444.68	22.49
	Mn <sup>2</sup> O <sup>3</sup>	989.37	30.32
	Mn O <sup>2</sup>	544.68	36.72
MERCY (Hydrargyrum)	Hg	1250.0	
Sulphide .....	Hg S	1150.75	13.84
MOLYBDENUM .....	Mo	575.83	
Bismuthido .....	Mo S <sup>3</sup>	977.33	41.08
Molybdic Acid .....	Mo O <sup>3</sup>	875.83	34.28
NICKEL .....	Ni	369.33	
Sulphides .....	Ni S	570.08	35.22
	Ni <sup>2</sup> S <sup>3</sup>	1340.91	44.92
	Ni S <sup>2</sup>	770.83	52.00
Oxide .....	Ni O	469.33	21.31
NITROGEN .....	Nb		
NITROGEN .....	N	175.0	
Nitric Acid .....	NO <sup>5</sup>	675.0	74.07
OSMIUM .....	Os	1242.62	
OXYGEN .....	O	100	
PALLADIUM .....	Pd	665.48	
PELOPIUM .....	Po		
PHOSPHORUS .....	P	392.28	
Phosphoric Acid .....	PO <sup>5</sup>	892.28	50.05
PLATINUM .....	Pt	1232.08	
POTASSIUM (Kalium) .....	K	489.30	
Potash .....	KO	589.30	16.97
RHODIUM .....	Rh	651.96	
RUTHENIUM .....	Ru	652	
SELENIUM .....	Se	495.28	
SILICIUM .....	Si	277.78	
Silica .....	Si O <sup>3</sup>	577.78	51.92
SILVER (Argentum) .....	Ag	1349.66	
Sulphide .....	Ag S	1550.41	12.95
Chloride .....	Ag Cl	1792.04	24.72
Iodide .....	Ag I	2935.66	54.02
Bromide .....	Ag Br	2349.28	42.55
Oxide .....	Ag O	1449.66	6.90

NAME OF SUBSTANCE.	FORMULA.	WEIGHT.	PER-CENTAGE. E-N. EL.
SODIUM (Natrium)	Na	289.73	
Soda .....	Na O	389.73	25.06
STRONTIUM.....	Sr	545.03	
Strontia .....	Sr O	645.03	15.48
SULPHUR.....	S	200.73	
Sulphuric Acid .....	SO <sub>3</sub>	500.73	50.91
TANTALUM (Ta, Pb, Nb)	Ta (Pb Nb)	2290.7	
Tantalic Acid (including pelopic and niobic acid).....	TaPbNbO <sub>3</sub>	2590.7	11.55
TELLURIUM.....	Te	801.76	
TERBIUM.....	Tb		
THORIUM .....	Th	743.80	
Thoria .....	Th O	843.80	
TIN (Stannum) .....	Sn	735.30	11.83
Binoxide.....	Sn O <sub>2</sub>	935.30	
TITANIUM .....	Ti	301.55	21.38
Sesquioxide.....	Ti <sub>2</sub> O <sub>3</sub>	903.10	33.22
Titanic Acid .....	Ti O <sub>2</sub>	501.55	39.88
TUNGSTENUM (Wolframium).....	W	1183.0	
Tungstic Acid .....	WO <sub>3</sub>	1483.0	20.23
URANIUM .....	U	750.0	
Oxide .....	UO	850.0	11.76
Sesquioxide.....	U <sub>2</sub> O <sub>3</sub>	1800.0	10.66
VANADIUM .....	V	856.89	
Vanadic Acid .....	VO <sub>3</sub>	1156.89	25.93
YTTRIUM.....	Y	402.5	
Yttria .....	YO	502.5	19.90
ZINC .....	Zn	406.59	
Sulphide .....	Zn S	607.34	33.05
Oxido .....	Zn O	506.59	19.74
ZIRCONIUM.....	Zr	420.20	
Zirconia .....	Zr <sub>2</sub> O <sub>3</sub>	1140.40	26.31





UNIVERSITY COLLEGE, TORONTO.

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Mineralogy and Geology.

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PROFESSOR CHAPMAN,

LATE PROFESSOR IN UNIVERSITY COLLEGE, LONDON,

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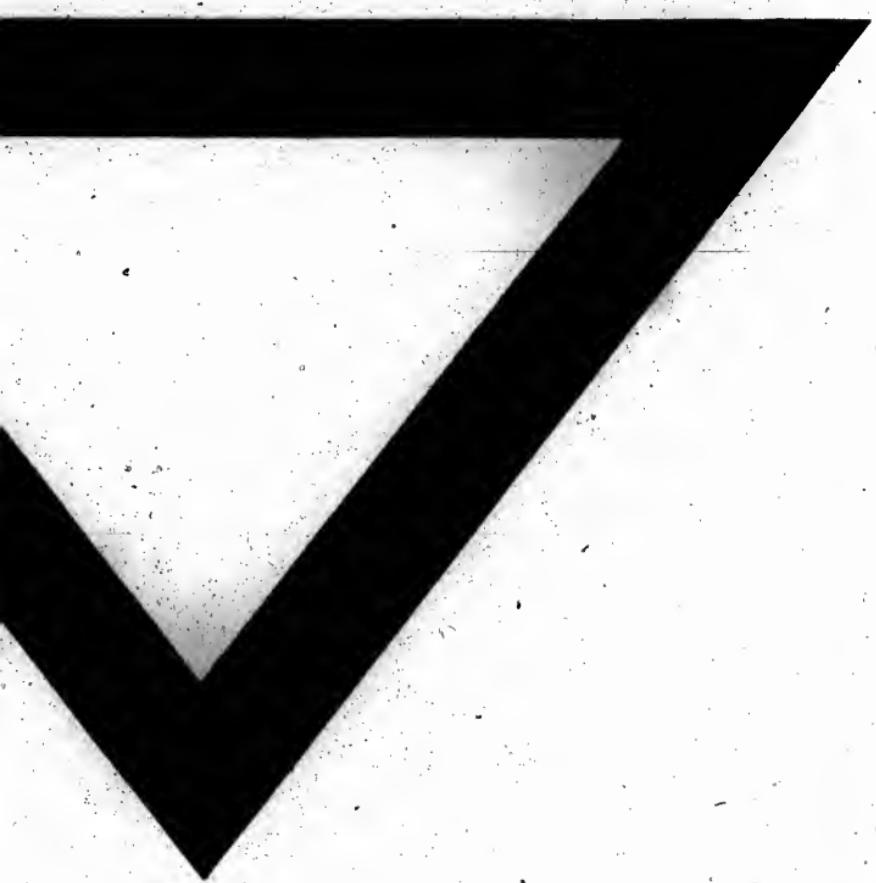
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