COMPOSITION OF LEAD ARSENATES.

Sample No.		Water	Lead Oxide	Arsenic Oxide	Figur dry b		Lead ate	
	Manufacturer	%	Pb0 %	As ₂ O ₈	As ₂ O ₅	Pb0 %	Рьн %	Pb _a %
347 348	Graselli Co., Cleveland. Chemical Laboratories.	40.49	40.335	16.744	28.14	67.78	44.00	53.00
349	TorontoSherwin-Williams,	42.81	41.285	13.886	24.28	72.19		100.00
306	Cleveland Niagara Brand Spray	50.48	35.020	13.678	27.62	70.72	29.66	69.84
372	Co., Burlington Hemingway London	24.23	54.492	18.544	24.47	71.92		98.00
012	Purple Co		32.904	16.634	32.18	63.67	92.50	6.50
373 436	Swifts	45.14	34.803	17.774	13.40	63.44	95.60	2.00
555	Eng E. Merck, Darmstadt,	47.24	30.564	16.429	31.14	57.93	100.00	
	Germany	1.01	73.425					
346	Graselli Co., Cleveland.	0.528	68.14	27.147	27.29	68.50	33.64	62.58

The 6th and 7th columns showing the composition when figured to dry basis illustrate how the per cent. of lead (PbO) and arsenic (As₂O₅) vary as the form of the lead arsenate varies. The higher the quantity of neutral arsenate the lower is the per cent. of arsenic and the higher the per cent of lead, and vice versa.

Lead arsenates should be of such a quality that when mixed with water they will give only inappreciable quantities of soluble arsenic, not more than 0.75 per cent. Samples which, when boiled with ammonia water, will give in the filtrate a precipitate with lead acetate and acetic acid, are as a rule injurious to foliage. The addition of a solution of lead acetate or nitrate to such arsenates till there is an excess of lead, as shown by potassium iodide paper (see under home-made lead arsenate), would be beneficial.

HOME-MADE LEAD ARSENATE.

If one cares to he can make his own lead arsenate at home. The only drawback to this is the assurity of obtaining materials of suitable quality. Bulletin 131, Bureau of Chemistry, U.S. Dept. of Agriculture, gives the following directions:

Formula A.												Ozs.
Sodium Arsenate (65%)				 							 	8
Lead acetate (sugar of	lead)										 	 22