

100 grains of harley meal give, on burning,	
ins of ash. 100 grains of this ash contains, of	
Silica,.....	29.67
Pho-phoric acid,.....	36.80
Sulphuric acid,.....	0 16
Chlorine,.....	0.15
Per-oxide Iron,.....	0.83
Lime,.....	3.23
Magnesia,.....	4 30
Potash,.....	16 00
Soda,.....	8.00

100.00

is analysis was made by PROF. THOMPSON Glasgow. German chemists have found some less than 3 per cent. of ash in 100 parts of barley. In a good soil adapted to the plant, quantity found by Dr. T. may be regarded as average. Supposing that all the straw was re- turned to the soil either directly, or in the shape of manure, 2000 lbs. of barley, after it was thoroughly dried at 212° taken from an acre, (equivalent to 50 bushels,) would remove from the earth 60 lbs. of its most valuable minerals. Among these would be 17.80 lbs. of soluble silica or flint, which is found mainly in the hull of barley. The most valuable earthy element in the seed of this grain, and in all others, is phosphoric acid. Of this, 2000 lbs. of barley remove from the soil 22 lbs. If barley is fed to swine, horses, or other domestic animals, and all their solid and liquid manure be returned back on the field that produced the crop, the soil will be made richer for the operation. This does not accrue from the mistaken idea that all the matter removed in a crop, can be returned back in the manure, which the crop will produce when fed to animals. All animals literally burn their food in their warmed bodies, the gases and vapor formed by the combustion escaping through the lungs in expired air. 100 lbs. of dry dung will make less than 50 lbs. of dry dung and 10 lbs. of urine. How, then, can the removal of 100 lbs. of barley, corn, or wheat, from a piece of ground, be the return of only 40 lbs. of the same matter to the soil? This question we greatly desire every child, whether male or female, 14 years of age should be able to answer correctly. We are disappointed and mortified at our poor success in persuading those that till the earth that schools to teach the laws of nature which govern the growth of cereals and other cultivated plants, ought to be established and supported for the benefit of agriculture. Let that pass; while we ask again why it is that a farmer may take 1000 lbs. of barley from a field and by restoring only one-tenth of the same, the soil of this land none the poorer by the operation? It will be recollected that 1000 lbs. of this grain contain at most but 30 lbs. of uncombustible earthy matter. By restoring these and 70 lbs. of organic matter, carbon, oxygen, hydrogen, and nitrogen, the husbandman gives back to the soil as much as he has furnished towards the 1000 lbs. of barley. How is the land made richer when we restore to the soil a particle more soluble flint, phosphorous, sul- phur, iron, lime, magnesia, soda, potash, chlorine, and nitrogen, that was removed in the crop?

Some knowledge of agricultural geology is indis- pensable to the clear understanding of this inter- esting subject. Consider for a moment the source from which all soils derive their lime, potash, phos- phorus, sulphur, and other earthy elements of plants. In 100 lbs. of the ash of barley there are 37 lbs. of phosphoric acid, 24 lbs. of potash and soda, beside 7½ lbs. of lime and magnesia. No wonder that this crop requires good land to yield a large amount of seed.

It is the constant abrasion, comminution, and so- lution of the small particles of rocks, which lie ex- posed to the meteoric influences of frost, heat, light, electricity, water, oxygen, carbonic and nitric acids from the air, that renovate soils while at rest, when partially exhausted by the removal of crops. Sci- ence can render the practical farmer most valuable aid in hastening the natural process for bringing back virgin fertility to a worn out field. But alas, the practical man too often scorns the proffered light of science. He ridicules the idea of having his sons study the properties and source of the constituent elements, which God has appointed to make the bread, the meat, and the clothing of all rational beings.—We rejoice however in the strong faith that this deep prejudice against the study of the natural sciences which have so intimate a connection with rural industry, can not last always. Our children's if not our own offspring, will see the end of it.

As a bushel of barley can be grown on some soils about as easily as a bushel of oats, who would not give a trifle to know by actual experiments the re- lative value of 200 lbs. of barley meal and a like weight of corn meal, for making pork, beef and mutton? By making meal into well cocked pud- ding, and mixing it with boiled or steamed potatoes, a little slop from the kitchen and dairy, pork can be made at no great expense, while the dung and urine from the pig sty will make great barley next year. We are much in favor of that system of husbandry which consumes the largest amount of the products of the farm at home, and carefully saves, and uses to the best advantage the manure thus made. We have often help to harvest from 45 to 50 bushels of barley on an acre, and have seen it much used in fattening hogs. But its precise value for feeding has never been determined.—*Gen. Eur.*

Green Peas for Winter Use.—The lovers of green peas will be pleased to learn that they can be preserved for winter use, by simply gathering them at the proper season for using them green, shelling them and drying them in the shade, and when well cured and perfectly dry, packing them away for use.

When required for use, they should first be im- mersed in warm water for ten or twelve hours, which will render them as tender and delicious succulent as when taken from the vines. The best method of preserving them, after they have been thoroughly cured by the above process, is to put them into close jars or bottles. In this way, not only green peas but green beans and green corn may be had the year round.—*Far. & Mesh.*