(8) Trimethyl Phosphite, C<sub>3</sub>H<sub>5</sub>O<sub>3</sub>P CA No. 121-45-9 HS No. 29.20.90.90 TDG No. 2329 NIOSH/RTECS No. TH 1400000

Synonyms: Methyl phosphite; phosphorous acid, trimethyl ester;
trimethoxyphosphine

**Physical Properties:** MW 124.09, mp: -78°, bp: 112°, d 1.052,  $n_D^{20}$  1.4080. Soluble in hexane, benzene, acetone, alcohol, ether and carbon tetrachloride; insoluble in water.

Synthesis: Phosphorous trichloride reacts with methanol in the presence of a base to form trimethyl phosphite.

Reactivity: Trialkyl phosphites react with carboxylic acids to form esters; at high temperatures these esters undergo an auto-Arbuzov rearrangement viz (CH<sub>3</sub>O)<sub>3</sub>P\_CH<sub>3</sub>P(O) (OCH<sub>3</sub>)<sub>2</sub>. Emits toxic fumes, PO<sub>4</sub>, on decomposition and is a flammable liquid.

**Toxicology:** Moderately toxic by ingestion and dermal contact, it is a severe skin and eye irritant. Its  $LD_{50}$  is 1600 mg/kg (rat) with a TLV of 2 ppm. It is reported in EPA TSCA inventory.

**Uses:** An intermediate in the production of persticides, fire retardants and organophosphorous additives. Also used in dyestuffs, optical brightners, plasticizers and lubricants.

Suppliers: There are 8 suppliers listed worldwide. These are Mexico (1), USA (2), PRC (2), Japan (1), Germany (1) and Switzerland (1); SRI International lists 5 USA producers.