

Mn(II) and Fe(III)] and carbon catalyze the oxidation of dissolved  $\text{SO}_2$  to form  $\text{H}_2\text{SO}_4$  in the absence of oxidants. The aqueous-phase may be acidified by the reaction of dissolved  $\text{HNO}_2$  with oxidants, but since limited data exists on ambient  $\text{HNO}_2$  concentration levels, the importance of this reaction is unknown. The relative importance of the gas-phase and the aqueous-phase pathways for various geographical regions has not been established; but there are indications that the aqueous-phase pathway is dominant for  $\text{H}_2\text{SO}_4$  formation. The relative importance of the pathways for  $\text{HNO}_3$  formation is not known, but it appears that the overall  $\text{HNO}_3$  formation rate is greater than 5 times that for  $\text{H}_2\text{SO}_4$ .

#### 4.3.1 The Photochemical Oxidation Cycle

Of the four major acidification (oxidation) pathways shown in Figure 4.1, the first three depend upon free radicals produced directly and indirectly in the photochemical oxidation cycle for polluted atmospheres. This cycle (Figure 4.2) is governed by the following basic features. Free radical attack on atmospheric volatile organic compounds (VOC) is initiated by a select group of compounds which are for the most part activated by sunlight. Formaldehyde and nitrous acid, in particular, show high potential as free radical initiators during the early morning sunrise period. After the initial free radical attack, the VOC's decompose by various paths resulting in the production of peroxy radical species ( $\text{HO}_2$ ,  $\text{RO}_2$ ,  $\text{R}'\text{O}_2$ , etc.) and partially oxidized products which in themselves may be photoactive radical-producing compounds (R is a fragment of an organic molecule). The peroxy radicals react with  $\text{NO}$ , converting it to  $\text{NO}_2$ , and in the process produce hydroxyl/alkoxyl radical species ( $\text{HO}$ ,  $\text{RO}$ ,  $\text{R}'\text{O}$ , etc.). Alkoxyl radicals can be further oxidized, forming additional peroxy radicals and partially oxidized products, thereby completing the inner cyclical loop reaction process illustrated in Figure 4.2; or they may attack the VOC pool (this is the major path for hydroxyl radical) present in the polluted atmosphere, thereby completing the outer loop