## APPENDIX.

I have been asked to desecribe mpparatus, solutions, ete., nealed to carry out the method of tenting for Chlorine.

The following are needed :

## Apraralus.

1 Burrette, 50 ce. graduated in 1.10 ..... 82.00
1 " stand. ..... 1.00
a dars funnel (2 inch.). ..... 0.10
ntirrer ..... 0.05
.orcelain dishes ( 4 inch. diam.) ..... 0.60
(White saucers will do.)
110 cc. pipette. ..... 0.20
150 cc. graduate tlask- ..... 0.20
84.15

## Reagents.

1. Distilled water.
2. Solution of Nitrate of silver, 4.788 i grammes of the crystals to 1 litre.

Each 1 cc . of this solution ecrresponds to 1 milligramme of Chlorine, (in chlorides.)
3. Chromate of Potash, a 5 per cent. solution.
4. Solution of Sodium Chloride, 1.648 grammes per litre.

Each 1 cc. contains 1 milligramme of Chlorine.
The method of working is so simple that it may easily be laarnt by any one having an elementary knowledge of chemistry, in a few hours. It is fully described ir. Jutton's Volumetric Analyais, (publithed by J. and A. Churchill, Iondon) and in other similar works. Beat, however, by personal instruction. hrazfrï, mern

Exceptional samples of water may require preliminery treatment; but these are so rarely met as to be negligible, or they may be sent to properly qualified analysts for an opinion.

