## PUZZLES EXPLAINED

there are two articles in general that have undoubtedly caused many persons to scratch their heads in perplexity—the automatic eigar-lighter and the thermos, or hot and water bottle.

cold water bottle. The pocket-lighter consists of a small tube, at the bottom of which is packed some cotton soaked in alcohol, ether or other inflammable and easily evaporated liquid, at the top of which evaporated liquid, at the top of which is a small framework holding a thin vertical wire. When the top of the tube is opened the wire in a short time becomes red-hot and the alcohol vapor ignited therefrom. The first explanation one would na-turently offer is that the friction of

The first explanation one would ha-turally offer is that the friction of the rapidly-moving vapor would render the wire warm; but on second thought this would seem to be too great an effect for so small an amount of energy.

The real reason ,however, lies in the wire itself. In most of the lighters the action is practically the same. The wire is either what is known as platiwhere is either what is known as plati-num sponge or some other metal sponge that has the power of absorb-ing gases rapidly. Whenever anything absorbs gases, that thing has made the gases occupy less space or volume, and whenever gases occupy less vol-ume they are said to be compressed, and whenever gases are compressed they invariably become hotter, the amount of this heat depending on the amount of compression. '

ABSORBERS OF CASES.

A great many things possess A great many things possess the power of absorbing gases, charcoal be-ing a notable example. Although char-ecoal absorbs a good many times its own volume of gases, it does not be-come appreciably hotter. It takes the metal sponge for that; for this absorbs hundreds of times its own volume of gases and does it in a short time, thus not allowing for the radiation of the generated heat. So the illtle wire is not allowing for the radiation of the generated heat. So the little wire is generally heated to redness, ignites the wick fed by the alcohol, and the thing is ready for business. To afford some idea of the intensity of heat necessary for this, it may be stated that to red heat a wire requires several hundred degrees. When a gas

several hundred degrees. When a gas is heated one degree centigrade it ex-bands exactly one two-hundred-and-seventy-third of its volume at zero centigrade. Conversely, when a gas is compressed one two-hundred and seventy-third of its volume at zero centi-grade, it is heated on degree. Now, suppose that a gas is compress-ed to one two-hundred and seventy-

third of its volume at zero centigrade, it will be heated to two hundred and seventy degrees, an intensity sufficient for our purpose. Thus, the spong-has to absorb only, let us say, one two-or three-hundredth of its own volume of the sufficient of the source of the second volume of the supplied gas. As to how it does this, or just what inherent roperty enables it to do it, is not Well !

HOT WATER CONTAINERS. In the case of the hot and cold water container, the makers claim that some varieties will maintain any liquid hot or cold for a period of 48 hours. In this we have the whole theory of the dissipation of heat: for cold simply means the absence of heat. There are three ways in which heat may be earried from one body to another, or, earned from one body to another, or, rather, imparted by one body to ano-ther-by conduction, by convection and by radiation. If all the precau-tions possible be taken to prevent the action of these three operations, little transmission of heat can take place; so the hot liquid does not lose its heat, and the cold liquid does not receive heat from the heated air or bodies out-

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ALE.



This fact is utilized in heating houses by hot water. The heated water rises through the pipes, gives out its heat to the room, and on becoming cooler sinks through the inlet pipe to 101

be heated again. It is the same with gases. Suppose that a bottle nuder consideration had a full volume of air between its lay-ers. If the inside of this bottle is hot ers. If the inside of this bottle is hot the inside of the air space immediate-ly becomes hot by conduction; this hot part rises, and the cooler air flows in to be heated likewise. But the heated air flows to the outside layer and heats it by conduction, and this in turn is delivered to the out-side air. But take all the air out and an chaver line can massible take place no convection can possibly take place. Thus is the second objection over-

turn becomes hot and rises.

The third and last one, that of ra-

The principle made use of is that of the De War flask, all the bottles on the market to-day being simply varia-tions of this piece of scientific appara-tus. The bottle is made to discount as far as possible the three operations mentioned. It consists of two layers of some polished metal with a space between them that has been exhausted of als as much as possible. The two of air as much as possible. The two layers are connected by just as few links as possible.

links as possible. Taking up the three methods of transmission in succession, we shall see how admirably each one of them has been rendered inactive. By con-duction of heat is meant the trans-mission of heat by a solid or some-thing acting as a solid in which the medium itself does not move as to relative situation of its parts. For instance in heating in a poker in a

relative situation of its parts. For instance, in heating in a poker in a fire, one end is thrust in and becomes redhot, and in a little while heat is conducted 'along' the iron until the opposite end becomes hot. The same applies to a teaspoon in a cup of hot tea or coffee. This is why the bottle referred to has as few connecting inks between layers as possible, so us to make the conduction of heat a minimum, considering, of course, a proper strength of the bottle as a whole.

The next method is that of convec-The next method is that of convec-tion. By this is meant the method ot transmitting heat by moving particles of a liquid or a gas. Take, for in-stance, the heating of water. The kettle is put over the fire, and the bottom layer of water is heated at once by the fire by conduction. But

beated liquid expands, and therefore becomes lighter than the cool liquid above; so it has to rise. It is then displaced by the cooler liquid, which

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