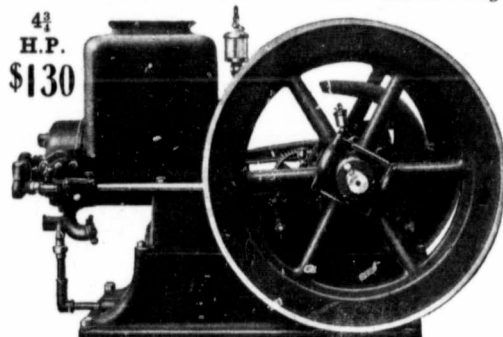
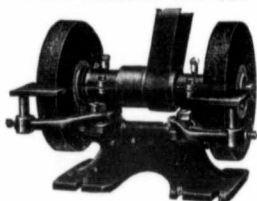


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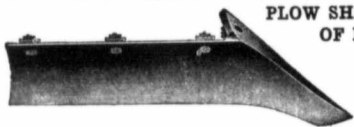
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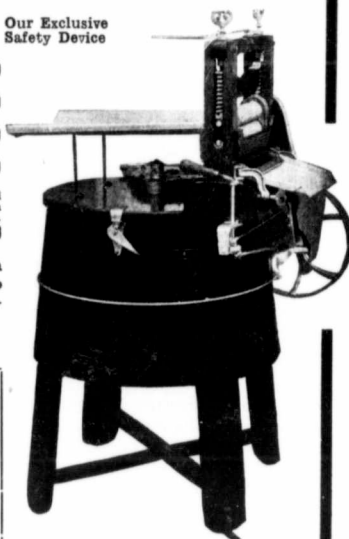
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The Canadian Stover Gasoline Engine Co. Ltd., Brandon, Man.**Gas Engine Troubles**

By A. C. CAMPBELL

Article Number 4

AT the close of our last article we were discussing the question of running an engine too cold and thereby losing power or using an excessive amount of fuel for the power delivered. To enable our readers to more thoroughly understand this matter it will be well to devote some time to the discussion of the relationship between heat and work. Heat is not a substance but a condition. It is a form of energy.

In measuring anything, whether a substance or a condition, we must have a definite unit. For instance, one pound is the unit of weight, one foot the unit of length and so on in all measurements. A degree of temperature is the unit used in measuring the intensity of heat but not the amount of heat. The unit of the amount of heat is called the British thermal unit and is invariably expressed by the letters B.T.U. A B.T.U. is the amount of heat required to raise the temperature of one pound of water,

one degree F when the temperature of the water is 62 degrees F.

There is a definite relationship between heat and work, that is to say, if we have a certain amount of heat in a fuel it is equivalent to a certain amount of energy or work.

Careful experiments have gone to prove that the mechanical energy of one B.T.U. is 778 foot-pounds. A pound of fuel contains a definite number of B.T.U. and therefore the greater number of heat units transformed into work by the engine, the more efficient is the engine. A pound of gasoline contains from 18,000 to 20,000 B.T.U. and a pound of kerosene contains from 22,000 to 24,000 B.T.U. If we conducted a test with a gasoline engine and found that of the 20,000 B.T.U. in a pound of gasoline, 5,000 were delivered in the form of work then the efficiency of the engine would be 25 per cent. That is to say, 25 per cent of the B.T.U. have been transformed into work. It should now be plain to the reader that if



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