ween protoxyd of roxyd of iron, and being as dichlorid, in the equation **A**. nner gives rise to irds dichlorid and ation **B**.

+ 2CuCl. 2Cu.

ee in B. is readily CuCl) of A, it folr be dinoxyd, the hloridized by the

e the protochlorid ble in water, but ind of most other 194° Fahrenheit n 16 per cent. of e than 8 per cent. red of water, dis-° F. (40° C.) 6·0 dichlorid of copvater they deposit der. A solution ater, dissolves at 4° F. (40° C.) bove figures are experiment. 100 per.

ight of protosul mmon salt 58.5,

31.7 9 while that of copper is 3.17, and that of iron 2 the proportions for the bath, numbers a little given, to allow for impurities in the salts empl three equivalents, say 95.1 pounds of copper, of protoxyd, two equivalents of protochlorid from 280 pounds of copperas and 120 pound 100 imperial gallons or 1000 pounds of water. convert one third of the copper into protochlor dichlorid, equal to 98.9 pounds of dichlorid of this soluble we have prescribed the addition of cent. additional of salt to the bath, while, as we holding but 15 per cent., will dissolve at 194° 1 than this quantity. While these are the theore is in practice, from a secondary re-action resul of an insoluble basic per-salt, a loss of protoch ing to from five to ten per cent., so that the c the bath is somewhat less than above represen considerable portion of dinoxyd of copper be p amount of dichlorid will be larger than above e of these reasons it is not well to add to the a 60 or 70 pounds of copper for each 100 galle the copper from a properly prepared ore wi by several hours digestion or percolation, eve more rapid in proportion as the heat approach

Preparation of the Ores.—In the calcination recommended, the object is to expel the carbon otherwise cause much effervescence in the bath dinoxyd predominates in the ore, this should be, verted into protoxyd by calcination in the air, leaves one third of its copper undissolved in the ing sulphuretted ores by roasting till all the there is obtained a mixture of sulphate of copp a portion of dinoxyd, which latter, according to F to 20 or 30 per cent. of the copper. Farther roa will convert both this and the sulphate into prodition is less advantageous, inasmuch as both