

TABLE VI.

Temp.	Concentration.	Spec. Conductivity.
25°C	Water	2.454×10^{-6}
"	$24.6 \times 10^{-7} \text{ n AgNO}_3$	2.842×10^{-6}
"	AgNO ₃ sol. along with $4.4 \times 10^{-8} \text{ n HCl}$	2.802×10^{-6}
"	$8.8 \times 10^{-8} \text{ n}$ "	2.892×10^{-6}
"	$17.7 \times 10^{-8} \text{ n}$ "	3.000×10^{-6}
"	$26.6 \times 10^{-8} \text{ n}$ "	3.033×10^{-6}
"	$88.8 \times 10^{-8} \text{ n}$ "	3.039×10^{-6}
"	$155.5 \times 10^{-8} \text{ n}$ "	3.046×10^{-6}
"	$244.4 \times 10^{-8} \text{ n}$ "	3.315×10^{-6}

TABLE VII.

Temp.	Concentration.	Spec. Conductivity.
25°C	Water	2.613×10^{-6}
"	$29.6 \times 10^{-7} \text{ n AgNO}_3$	3.201×10^{-6}
"	AgNO ₃ sol. along with $4.4 \times 10^{-8} \text{ n HCl}$	3.182×10^{-6}
"	$8.8 \times 10^{-8} \text{ n}$ "	3.189×10^{-6}
"	$17.7 \times 10^{-8} \text{ n}$ "	3.306×10^{-6}
"	$44.4 \times 10^{-8} \text{ n}$ "	3.521×10^{-6}
"	$88.8 \times 10^{-8} \text{ n}$ "	3.600×10^{-6}
"	$155.5 \times 10^{-8} \text{ n}$ "	3.715×10^{-6}
"	$244.4 \times 10^{-8} \text{ n}$ "	2.050×10^{-6}