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## OXYMETHYLENE AND FORMYL COMPOUNDS

GROUP IV..... S - C = Oand Group V..... -N - C = N - C = 0and -N=C-N=Ċ-CLASS II......GROUP I...... O-C=CMixed and termin. and 0=Ċ Mixed and terminal carbon atoms. GROUP II... - N - C = Cand  $-N = \dot{C}$ GROUP III..... H-S-C=C-S=Ċ and PERISSAD ..... O-C=C-C=Nand 0=Ċ CLASS III.......GROUP I......--C-C=C-Unsaturated hy. and drocarbon chain. CLASS IV......GROUP I... O - N = N and O = N - NΉ. Chains without carbon. GROUP II..... -N - N = N -|H and -N = N - N -CLASS V.....GROUP I...... O-N=Cand O = N -Middle nitrogen and terminal carbon atoms. GROUP II..... -N-N=Cand  $-\dot{C}=N$ and Two carbon atoms bound together by nitrogen.

The oxymethylene compounds belong to Laar's second class of triads, in which besides a middle carbon atom, one has at one end another atom of the same kind. To this class belong the ketones, aldehydes and the phenols, which may be represented by the tautomeric groups.



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