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## Farm Yard Manure.

aw subjects deserve greater consideration at hands of Canadian farmers than the advanous management and application of farm Many thousands of pounds are ally lost from ignorance or neglect of these Professor Voeleker, the rtant matters. alting chemist of the Royal Agricultural ty of England, a short time since went into laborious investigations respecting the osition of this kind of manure, and the ges and deterioration to which it is ordi-"subjected. His experiments were made considerable quantities, treated in different ers, for the purpose of observing the ges it undergoes; and great care appears ve been exercised in all the operations, articularly in selecting samples for analysis. these investigations it would appear that reat loss to which farm yard manure is this not so much from evaporation of aminto the air, as has hitherto been comsupposed, but chiefly from the action of washing out the soluble matter. fammonia by exposure, unless the mass is ted to a considerable degree of fermentappears to be in reality very small. ing table, giving the composition in pounds experimental heap of manure at four dif-

ferent periods, will afford the reader some idea of the changes which occur:—

	Put un Nov. ; J, 1854.	April 30, 1855.	August 23rd.	Novem- ber 15.
Weight of manure in the Water	2838. 1877.9	2026. 1336-1	1.94. 1505 3	1974. 1466.5
Dry Matte	960.1	689.9	488.7	507.5
Soluble organic matter.		ზნ.51	58.83	54.04
Soluble inorganic mat-	4371	57 88	39.16	36.89
Insoluble organic mat-	731.07	389.74	243.22	214,92
Insoluble inorganic matter	114.94	155 77	147.49	201.65
Total nitrogen Equal to ammonia	960 1 15.23 22.14	689.9 18.14 22.04	488.7 13.14 15.96	507.5 13.03 15.75

It is to be observed that, during the first six months, although the weight of the manure largely diminished, the loss was almost exclusively confined to the insoluble organic matter; while the soluble matter had increased, and the ammonia remained undiminished. But during the hot summer weather all the most valuable matters had undergone diminution.

Many important and elaborate analyses made by Dr. Vooleker, show the composition of the dung when treated under different systems. The conclusions to which they lead are these:— Farm yard manure in its fresh state contains but a small quantity of ammonia, most of its nitrogen being there as insoluble nitrogenous matter. But as the decomposition advances the ammonia increases, and a quantity of organic matters become soluble. For this reason the manure