PART II

LXXVII.—Some Esters of Antimony Trioxide.

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In a previous paper, Lang, MacKey, and Gortner (Trans., 1908, 93, 1364) described a method for the esterification of arsenious oxide with the fatty alcohols and with phenol and its homologues by removing the water produced by the reaction:

 $6ROH + As_2O_3 = 2R_3AsO_3 + 3H_2O$

as quickly as it was formed by means of anhydrous copper sulphate placed in a Soxhlet tubo attached to a flask containing weighed quantities of the reacting substances. By this means a large number of esters, which other methods had failed to produce, were prepared in quantity, and found to correspond with the general formula R_SAsO₃. In the present work, attempts were made to form the corresponding esters of antimony by five different methods, namely: (1) heating a mixture of the alcohol and antimony trioxide in a flask to which a reflux condenser was