

WYETH'S

GLYCEROLE CHLORIDE OF IRON.

(NON-ALCOHOLIC.)

Each fluid ounce represents twenty-four minims Tincture Chloride of Iron.

PALATABLE AND EFFICIENT.

Submitted to the usual *Chemical Test*, it develops the reaction characteristic of a *True Ferric Salt*, as contrasted with a reduced (Ferrous) salt.

Moreover, while retaining all the virtues of the Tincture of Iron Chloride, so essential in many cases, which no other salt of Iron (the Hydrochloric Acids itself being most valuable) can be substituted to insure the results desired, it is absolutely free from the objections hitherto urged against that medicament, being non-irritant, and will deservedly occupy an important therapeutic position in cases where Iron is indicated. It has no hurtful action upon the enamel of the teeth, even after long exposure.

In offering to the Medical Profession the above-described elegant pharmaceutical product, a word may be added in relation to the wide range of application to which it is adapted. Its preparation has received the greatest care in our laboratory, having aimed to produce a preparation that would be efficacious as well as pleasant—in the treatment of diseases to which this form of Iron is adapted—and in this desire, we have fully realized our most sanguine expectations.

Whether viewed from a *Therapeutical or Chemical standpoint*, it will be found indefinitely superior to any product of this character now in the market, and we earnestly desire the profession to submit it to the *test of clinical experience*.

CHEMICAL AND PHARMACEUTICAL COMPATIBILITY.—Glycerole, Chloride of Iron (Wyeth's) is compatible with the following *Alkaloids*:—*Quinine, Atropide, Godiene, Strychnine, and Caffeine*. It is also compatible with the following *Salts and Preparations*:—*Morphine Hydrochlorate, Ammonium Hydrochlorate, Cocaine Hydrochlorate, Potassium Hydrochlorate, Antifebrin, Antipyrine, Phenacetin, Salicin, Santonin, etc.* Hydrochlorates and Nitrates can be added only in small proportions; Sulphates and Acetates not at all. It is freely mixible in water in all proportions.

DOSE.—*Glycerole Chloride of Iron (Wyeth's)* may be given in doses ranging from a teaspoonful to a tablespoonful, an ordinary teaspoonful representing 8 to 10 drops (approximately 4 to 5 minims of the tincture.)

TEST.—Ferricyanide of Potassium in solution added to a small portion of *Glycerole Chloride of Iron (Wyeth's)*, changes it from a reddish-yellow to a green color (Test for Ferric salt), while the same added to a Ferrous salt results in a blue reaction. Preparations of Iron are more or less affected by exposure to the direct rays of sunlight, a fact which must be borne in mind concerning this product.

Price per dozen 16 oz. Bottles, \$9.00.

Our Agents shall be pleased to send samples of this new Iron preparation to physicians on request.

JOHN WYETH & BROTHER,

DAVIS & LAWRENCE CO., (Ltd.), Montreal.

GENERAL AGENTS.