regular octahedron on the other, will result.* These forms, moreover, as well as their intermediate oscillations, frequently occur in the same substance: red oxide of copper may be cited as an example. But between the cube, a square prism, and a rhombic prism, no relations of this ki. I exist. Neither are these forms related physically: for their optical, thermal, and other physical relations are equally distinct. By considerations of this sort, therefore, we are able to establish six (or really seven) distinct Crystal Systems. These (named chiefly in accordance with the relations of their axes) are enumerated in the annexed tabular view.†

Crystal-axes of one \(\) The Monometric System (including the cube, rhombic dodecahedron, octahedron, &c., length. Refraction. single with their various combinations. The Dimetric System (including square-Crystal-axes of two based prisms and pyramids with their lengths. Refraction, various combinations.) double, with one The Hexagonal System (including regular neutral line or ophexagonal prisms and pyramids, rhomtical axis behedrons, &c., with their combinations.) The Trimetric System (including right rectangular prisms and pyramids, Axes at rightangles. rhombic prisms Crystal-axes of three pyramids, and combilengths. Refraction, nations of these. The Monoclinic System double, with two One axis ob-(including oblique reeneutral lines or oplique. tical axes. tangular and rhombic combinations.) All the axes) The Triclinic System (including doubly-oblique oblique. combinations.)

^{*} The Law of Symmetry, in its exact acceptation, may be thus expressed:

If an edge or angle of a crystal be modified, all the similar edges or angles must be equally modified.

Or (2.) One-half or one-mth of the corresponding angles or edges, in alternate positions, must be equally modified. Example.—Oute and Tetrahedron (Boracite; Arseniate of Iron.) Or (3.) All the similar edges or angles must is modified by one-half or one-mth the normal or regular number of planes. Example.—Cube and Pentagonal Dodecahedron (Iron Pyrites.)

Conditions 2 and 3 produce hemihedrons or part-forms.

t See also Vol. V. of this Journal (New Series), pages 7-0.