

The experimental values  $\left(\frac{d\rho}{dp}\right)_{\text{exp.}}$  agree very closely with those for  $\left(\frac{d\rho}{dp}\right)_A$ , the greatest deviation being 2.8 percent and the least 0.9 percent. On the other hand, the experimental values differ from  $\left(\frac{d\rho}{dp}\right)_B$  by from 8.5 percent to 7.5 percent.

#### The Effect of Errors in Data

The following table shows the percent effect on the values at the heads of the columns caused by a one percent increase in the data shown in the first column:

	$\left(\frac{d\rho}{dp}\right)_A$	$\left(\frac{d\rho}{dp}\right)_B$	$\left(\frac{d\rho}{dp}\right)_{\text{exp.}}$
$\lambda_1$	0	0	+2
$\lambda$	0	0	-2
$\alpha$	+1.04 to +0.96	+1.15 to +1.17	0
$c_v$	+0.013 to +0.010	+0.15 to +0.17	0
$-h$	+0.05 to +0.09	0	0

Thus, to bring the values  $\left(\frac{d\rho}{dp}\right)_{\text{exp.}}$  down to the values  $\left(\frac{d\rho}{dp}\right)_B$  there would have to be consistent errors of about +4 percent in  $\lambda$ , or -4 percent in  $\lambda_1$  (or +2 percent in  $\lambda$  and -2 percent in  $\lambda_1$ ). Chance errors in the measurement of wave length are much less than this (see values, page 449), and it is quite improbable that methodical errors would have opposite signs in  $\lambda$  and  $\lambda_1$ .

To bring the values  $\left(\frac{d\rho}{dp}\right)_B$  up to the value  $\left(\frac{d\rho}{dp}\right)_{\text{exp.}}$  an error of about +7 percent would be necessary in  $d$ , or an error of about +50 percent in  $c_v$ . Errors of these magnitudes are practically impossible in  $d$ , and very unlikely even in the case of the doubtful  $c_v$ .